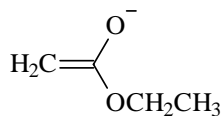


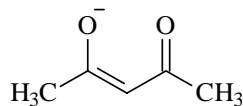
1. Rank the compounds below in order of decreasing basicity.



A

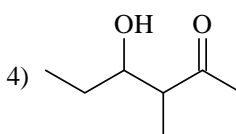
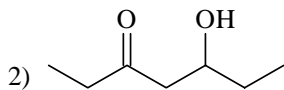
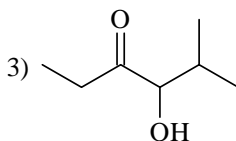
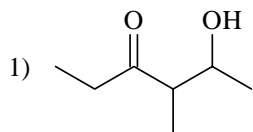
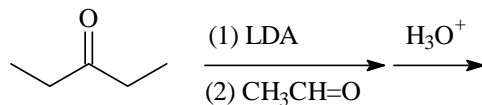


B



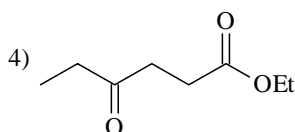
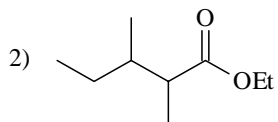
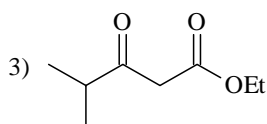
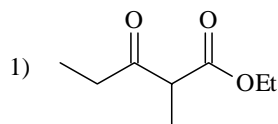
C

- 1) B>C>A 2) B>A>C 3) C>A>B 4) A>C>B
2. Which one of the following can give no Claisen condensation product?
- 1) $(\text{CH}_3)_3\text{CCO}_2\text{Et}$ 2) $\text{C}_6\text{H}_5\text{CH}_2\text{CO}_2\text{Et}$
 3) $\text{H}_2\text{C}=\text{CHCH}_2\text{CH}_2\text{CO}_2\text{Et}$ 4) $(\text{CH}_3)_2\text{CHCH}_2\text{CO}_2\text{Et}$
3. Which of the following is the product in the reaction shown below?



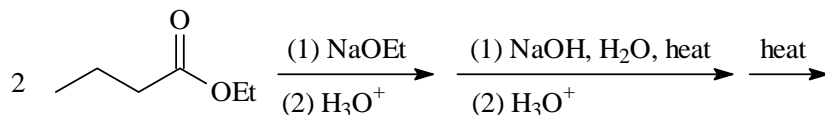
- 1) 1 2) 2 3) 3 4) 4

4. Which of the following is the Claisen condensation product of ethyl propanoate, $\text{CH}_3\text{CH}_2\text{CO}_2\text{Et}$?



- 1) 1 2) 2 3) 3 4) 4

5. Which of the following is the product of the reaction sequence shown below?



- 1) 4-methyl-3-hexanone 2) 4-heptanone
3) 2-propylbutanoic acid 4) 2-ethylpentanoic acid

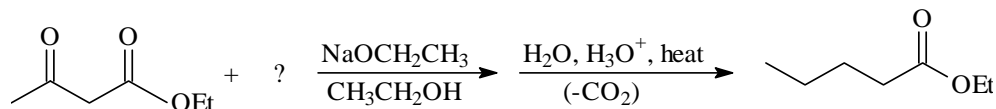
6. Heating butylmalonic acid, $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}(\text{CO}_2\text{H})_2$, to 140°C yields:

- 1) hexanoic acid 2) pentanoic acid
3) 2-methylpentanoic acid 4) 2-hexenoic acid

7. How many different Claisen condensation products are possible in the reaction of equal amounts of ethyl acetate ($\text{CH}_3\text{CO}_2\text{Et}$) and ethyl propanoate ($\text{CH}_3\text{CH}_2\text{CO}_2\text{Et}$)?

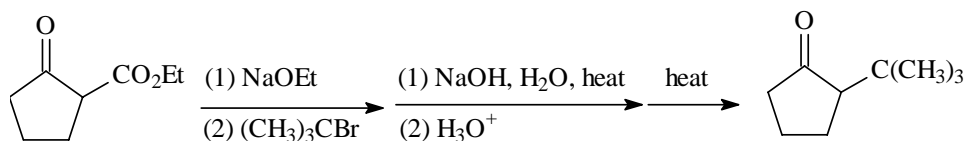
- 1) only one 2) two 3) three 4) four

8. Which of the following could be used as the missing reagent to carry out the following transformation?

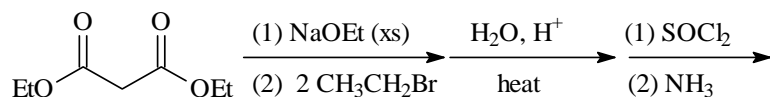


- 1) $\text{CH}_3\text{CH}=\text{O}$ 2) $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$
3) $\text{CH}_3\text{CH}_2\text{CH}_2\text{Br}$ 4) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{I}$

9. Consider the following synthetic scheme below. Which of the following best explains why the synthesis does not work?

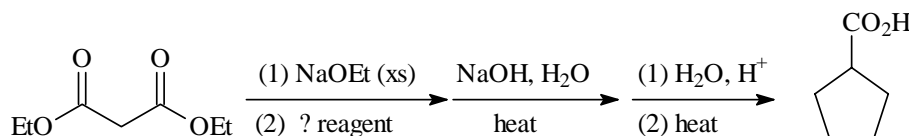


- 1) Using NaOEt gives Claisen condensation instead of alkylation.
 - 2) The alkyl halide used will lead to elimination rather than alkylation.
 - 3) The keto-acid formed does not decarboxylate in the last step.
 - 4) The base-promoted hydrolysis step does not work on the -keto ester intermediate.
10. What is the product of the following reaction sequence?



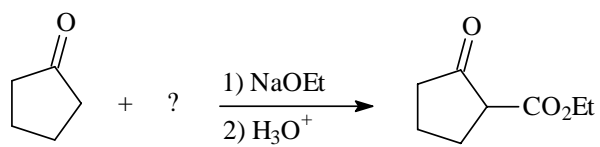
- 1)
 - 2)
 - 3)
 - 4)
- 1) 1 2) 2 3) 3 4) 4

11. What is the missing reagent in the synthesis shown below?



- 1) bromocyclopentane
- 2) 1,4-dibromobutane
- 3) 1,5-dibromopentane
- 4) 1,1-dibromocyclopentane

12. Identify the missing reagent in the reaction shown below.



- 1) ethyl formate, HCO₂Et
- 2) diethyl carbonate, (EtO)₂C=O
- 3) diethyl oxalate, EtO₂CCO₂Et
- 4) ethyl acetate, CH₃CO₂Et