Chemistry 211 Name _____
Exam 2 - March 17, 2004

15 1. Starting with an aldehyde or ketone show the reaction needed to produce the following compounds, using any other appropriate reagents:

2-hexanol

butanoic acid

Show the structure for each of the following compounds:

hemiacetal between acetaldehyde and ethanol

2-butene-2-ol (an enol)

4-cyano-4-heptanol (a nitrile)

9 2. Give names (IUPAC or common) of the alcohol, aldehyde and acid having the number carbons shown. If more than one compound is possible, list only one.

	Alcohol	Aldehyde	Acid
3-carbon			
4-carbon			
5-carbon			

12 3. (i) Starting with acetone, show the reaction mechanism (with arrows) required to produce an enol in acidic conditions. (ii) Using the enolate ion of acetone, show the reaction mechanism (with arrows) for an aldol condensation (product contains an alcohol and keto groups) using benzaldehyde, producing the same product we did in the lab. (iii) Show the α,β-unsaturated carbonyl-containing product from step iii, after the heating of this aldol product.

6 4. Draw structures for each of the following pairs of compounds. Circle the name of the compound in each pair which would be the most acidic (lower pK_a).

A. Fluoroacetic acid Trifluoroacetic acid

B. Ethanol Phenol

6 5. Starting with acetone (propanone) show the product for reaction between the enol tautomer and molecular bromine. Show the major product(s) for this reaction.

12 6. Show structures for the following compounds

benzoic acid

sodium butyrate

methyl propyl ketone

malonic acid

4-carbon lactone (cyclic ester)

Methyl benzoate

8 7. Show structures for each of the following functional groups or reaction products.

Alcohol

Ether

enolate ion

Aldehyde

Ketone

ester

nitrile

cyanohydrin

12 8. In lecture, we discussed the 6-step process to make an ester from a carboxylic acid and an alcohol. This reaction process produces a tetrahedral intermediate which is common to virtually all reactions involving carboxylic acids. Show this 6-step process, starting with acetic acid and methanol to produce methyl acetate. For each step below, show the product of the previous step as the reactant of the next step (product of step #1 becomes reactant of step #2, and so on).

1.

2.

3.

4.

5.

6.