

Astronomy Ranking Task: Chemical Reactions

Exercise #1

Description: The following are all unbalanced chemical reaction equations.

- A. $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- B. $\text{C}_4\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- C. $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- D. $\text{C}_8\text{H}_{18} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

A. Ranking instructions: Rank the number of oxygen atoms on the left side of each unbalanced reaction.

Ranking Order: Most 1 _____ 2 _____ 3 _____ 4 _____ Least

Or, all of the unbalanced reactions have the same number of oxygen atoms on the left side.
_____ (indicate with a check mark)

Carefully explain your reasoning for ranking this way:

B. Ranking instructions: Rank the number of oxygen atoms on the right side of each unbalanced reaction.

Ranking Order: Most 1 _____ 2 _____ 3 _____ 4 _____ Least

Or, all of the unbalanced reactions have the same number of oxygen atoms on the right side.
_____ (indicate with a check mark)

Carefully explain your reasoning for ranking this way:

C. Ranking instructions: Rank the number of carbon atoms on the left side of each unbalanced reaction.

Ranking Order: Most 1 _____ 2 _____ 3 _____ 4 _____ Least

Or, all of the unbalanced reactions have the same number of carbon atoms on the left side.
_____ (indicate with a check mark)

Carefully explain your reasoning for ranking this way:

D. Ranking instructions: Rank the number of carbon atoms on the right side of each unbalanced reaction.

Ranking Order: Most 1 _____ 2 _____ 3 _____ 4 _____ Least

Or, all of the unbalanced reactions have the same number of carbon atoms on the right side.
_____ (indicate with a check mark)

Carefully explain your reasoning for ranking this way:
