## Astronomy Ranking Task:

## Nuclear Physics

## Exercise \#1

Description: The following table has some information for three neutral atoms on the periodic table, labeled X, Y, and Z

| ranking letter | $\mathbf{X}$ | $\mathbf{Y}$ | $\mathbf{Z}$ |
| :--- | :---: | :---: | :---: |
| element symbol | B |  |  |
| proton number |  | 9 |  |
| neutron number |  | 10 |  |
| electron number |  |  | 18 |
| mass number | 11 |  | 40 |

A. Ranking instructions: Rank the numbers of protons in each atom.

Ranking Order: Least 1 $\qquad$ 2 $\qquad$ 3 $\qquad$ Most

Or, all of the atoms have the same number of protons. $\qquad$ (indicate with a check mark)

Carefully explain your reasoning for ranking this way:
B. Ranking instructions: Rank the numbers of neutrons in each atom.

Ranking Order: Least 1 $\qquad$ 2 $\qquad$ 3 $\qquad$ Most

Or, all of the atoms have the same number of neutrons. $\qquad$ (indicate with a check mark)

Carefully explain your reasoning for ranking this way:

## C. Ranking instructions: Rank the numbers of electrons in each atom.

Ranking Order: Least 1 $\qquad$ 2 $\qquad$ 3 $\qquad$ Most

Or, all of the atoms have the same number of electrons. $\qquad$ (indicate with a check mark)

Carefully explain your reasoning for ranking this way:
D. Ranking instructions: Rank the mass numbers of each atom.

Ranking Order: Least 1 $\qquad$ 2 $\qquad$ 3 $\qquad$ Most

Carefully explain your reasoning for ranking this way:

