## Astronomy Ranking Task:

## Nuclear Physics

## Exercise \#2

Description: The following table has some information for five atoms on the periodic table, labeled V through Z . Note that some may be ions.

| ranking letter | V | W | $\mathbf{X}$ | $\mathbf{Y}$ | $\mathbf{Z}$ |
| :--- | :---: | :---: | :---: | :---: | :---: |
| element symbol | Al |  |  |  | Na |
| proton number |  | 16 | 36 | 17 |  |
| neutron number |  | 16 | 50 |  | 12 |
| electron number |  | 18 |  |  | 10 |
| mass number | 27 |  |  | 35 |  |
| electrical charge | +3 |  | 0 | -1 |  |

A. Ranking instructions: Rank the numbers of protons in each atom.

Ranking Order: Least 1 $\qquad$ 2 $\qquad$ 3 $\qquad$ 4 $\qquad$ 5 $\qquad$ Most

Carefully explain your reasoning for ranking this way:
B. Ranking instructions: Rank the numbers of neutrons in each atom.

Ranking Order: Least 1 $\qquad$ 2 $\qquad$ 3 $\qquad$ 4 $\qquad$ 5 $\qquad$ Most

Carefully explain your reasoning for ranking this way:

## C. Ranking instructions: Rank the numbers of electrons in each atom.

Ranking Order: Least 1 $\qquad$ 2 $\qquad$ 3 $\qquad$ 4 $\qquad$ 5 $\qquad$ Most

Carefully explain your reasoning for ranking this way:
D. Ranking instructions: Rank the mass numbers of each atom.

Ranking Order: Least 1 $\qquad$ 2 $\qquad$ 3 $\qquad$ 4 $\qquad$ 5 $\qquad$ Most

Carefully explain your reasoning for ranking this way:
E. Ranking instructions: Rank the values of the electrical charges on each atom.

Ranking Order: Most Negative 1 $\qquad$ 2 $\qquad$ 3 $\qquad$ 4 $\qquad$ 5 $\qquad$ Most Positive

Or, all of the atoms have the same electrical charge. $\qquad$ (indicate with a check mark)

Or, there is not enough information to determine this. $\qquad$ (indicate with a check mark) Carefully explain your reasoning for ranking this way:

